Quiz (Occurrence and Discovery of Metals)

- 1. *P*, *Q* and *R* are three different metals. When each of the ores of these metals is heated strongly in air, only the ore of *Q* gives a solid with metallic lustre. When ores of *P* and *R* are heated with powdered carbon respectively, the ore of *R* gives a solid with metallic lustre. It is found that, *P* can only be extracted from its molten ore by electrolysis.
 - (a) Arrange the metals in order of decreasing ease of extraction from their ores, i.e. the easiest first.
 - (b) Deduce the order of discovery of the metals, the earliest first. Explain your answer briefly.
- 2. The table below lists some information about iron and gold:

Metal	Year of discovery	Relativeabundance (%)
Iron	3000 B.C.	5.0
Gold	5000 B.C.	0.000004

Explain the following statements:

- (a) Gold is rare but it was discovered much earlier than iron.
- (b) Gold is expensive although it exists as free element in nature.
- 3. Aluminium is commonly used to make soft drink cans. Used aluminium cans are usually collected for recycling.
 - (a) Explain why aluminium is usually used to make soft drink cans.
 - (b) What is meant by 'recycling metals'?
 - (c) Aluminium is the most abundant metal in the Earth's crust. However, it is still important to recycle aluminium. Give TWO reasons why aluminium is recycled.

Suggested Answer

- 1. (a) Q, R, P
 - (b) Q, R, P. The more easily a metal can be extracted, the earlier it is discovered.
- 2. (a) Gold is less reactive than iron. It can be easily obtained by physical method.
 - (b) This is because gold is very rare.
- 3. (a) This is because aluminium has low density, non-toxic, very malleable and corrosion resistant.
 - (b) Recycling metals means melting down used metals and using them again.
 - (c) Any TWO of the following:
 - It saves metal resources.
 - It saves energy and other resources such as electricity, water and fuels.
 - It reduces metal waste and land used for waste disposal.
 - It reduces pollution arising from the mining and extraction of metals.